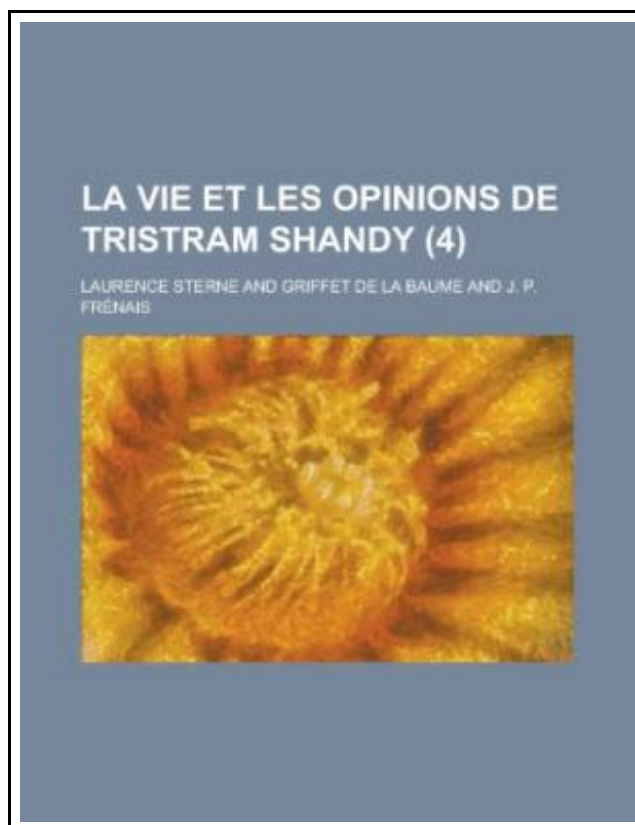


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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 42 pages. Original publisher: Hampton, Va. : National Aeronautics and Space Administration, Langley Research Center, 1985 OCLC Number: (OCoLC)166902673 Subject: Polyimides. Excerpt: . . . 14 contained the bulky-C (CF3) 2 group in the dianhydride. Although meta-isomerism of the aromatic diamine enhanced overall solubility of the polymer, use of this technique by itself did not produce soluble polymers. For example, the BTDA and PMA polymers in Table IV which contained meta-oriented APB and DDSO2 diamines were totally insoluble in all solvents tested. Likewise the incorporation of 3, 3 -diaminobezophenone, 3, 3 -methy-lenedianiline and m-phenylenediamine into both RTDA and PMDA polymers has produced insoluble films. 8 The increased solubility of polymers in this and previous studies 8 can be explained in terms of the variation of molecular structure to reduce electronic interactions or charge transfer complex formation. Linking groups in the dianhydride which serve to reduce electron affinity, bulky-CF3 and DDSO2 groups, and meta-oriented diamines all serve to reduce CTC. Reduction or elimination of CTC should greatly facilitate the solvation of polymer chains. In addition to the changes in polymer properties observed herein, preliminary data in our labs show that changing the molecular structure to reduce electronic interactions between chains in aromatic polyimides may affect not only film transparency and solubility, but may also enhance flow properties to provide improved processability. Effect of UV Irradiation on Film Color. The IIV-visible spectra of the more transparent films produced in this study (Table I) are reproduced in Figures 7 and 8. Transmission spectra of all of the 6F dianhydride-contain-ing films are compared in Figure 7 to that of commercial Kapton H film. The 0. 5 mil films were approximately g5 transparent at 500 nm, the wavelength of...

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